metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Andreas Decken

Department of Chemistry, University of New Brunswick, Fredericton, NB, PO Box 45222, E3B 6E2, Canada

Correspondence e-mail: adecken@unb.ca

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å Disorder in main residue R factor = 0.035 wR factor = 0.092 Data-to-parameter ratio = 10.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

$(\eta^5$ -Cyclopentadienyl) $(\eta^6$ -9*H*-fluorene)iron(II) hexafluorophosphate

In the title compound, $[Fe(C_5H_5)(C_{13}H_{10})]PF_6$, the fluorene skeleton is disordered over two positions and parallel with the cyclopentadienyl ring.

Comment

The title compound, (I), is an important starting material, employed in the synthesis of the contentious complex (η^{5} -cyclopentadienyl)(η^{5} -fluorenyl)iron (Johnson & Treichel, 1977; Ustynyuk *et al.*, 1986).



The cyclopentadienyl and aryl rings are parallel [dihedral angle = $1.9 (3)^{\circ}$] and the metal atom is coordinated centrally to both rings. This is in contrast to $(\eta^5$ -cyclopentadienyl) $(\eta^6$ fluorenyl)iron (Johnson & Treichel, 1977) where the Fe atom is displaced away from the ring junction between the five- and six-membered rings, leading to Fe-C distances ranging from 2.032 (5) to 2.316 (5) Å. In (I), the Fe-C distances are in the range 2.078 (14)-2.132 (6) Å, similar to the bonding pattern observed in the analogous 4H-cyclopenta[def]phenanthrene (Decken et al., 1997) and endo-9-(2-methylphenylamino)fluorene (Pierre et al., 1998) complexes. The fluorene skeleton is disordered over two positions as previously observed in the corresponding methyl-substituted complex (η^5 -cyclopentadienyl)-*exo*-(η^6 -9-methyl-9*H*-fluorene)iron hexafluorophosphate (Johnson & Treichel, 1977), where publishable data could not be obtained.



Figure 1

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved A view of the cation, with displacement ellipsoids drawn at the 30% probability level. Only the major component of the disordered fluorene group is shown.

Received 29 October 2004 Accepted 2 November 2004 Online 6 November 2004

Experimental

The title compound was prepared according to a published procedure (Nesmeyanov *et al.*, 1966).

 $D_x = 1.728 \text{ Mg m}^{-3}$

Cell parameters from 6469

Irregular fragment, orange $0.18 \times 0.13 \times 0.13$ mm

Mo $K\alpha$ radiation

reflections

 $\begin{array}{l} \theta = 2.3 \text{--} 32.2^{\circ} \\ \mu = 1.07 \ \text{mm}^{-1} \end{array}$

T = 293 (2) K

Crystal data

 $[Fe(C_5H_5)(C_{13}H_{10})]PF_6$ $M_r = 432.12$ Monoclinic, P_{2_4}/n a = 7.3373 (4) Å b = 9.5498 (5) Å c = 23.9163 (12) Å $\beta = 97.498$ (1)° V = 1661.48 (15) Å³ Z = 4

Data collection

Bruker AXS SMART1000/P4	3826 independent reflections
diffractometer	2711 reflections with $I > 2\sigma(I)$
ω and φ scans	$R_{\rm int} = 0.022$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.5^{\circ}$
(SADABS; Sheldrick, 1997a)	$h = -9 \rightarrow 9$
$T_{\min} = 0.780, \ T_{\max} = 0.875$	$k = -11 \rightarrow 12$
13243 measured reflections	$l = -31 \rightarrow 30$
Refinement	
$\mathbf{P} \in \mathbf{P}^2$	TT / / / / /

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.035$ $wR(F^2) = 0.092$ S = 0.973826 reflections 352 parameters H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0533P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.002$ $\Delta\rho_{max} = 0.37 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.31 \text{ e} \text{ Å}^{-3}$

H atoms were placed in calculated positions, with C–H distances fixed at 0.93 (CH groups) or 0.96 Å (CH₂ groups), and $U_{\rm iso}$ values set at $1.2U_{\rm eq}$ of the carrier C atom. The fluorene ligand was disordered

over two positions. The atomic distances in the minor component were restrained to the corresponding distances in the major component. The site occupancies were determined using an isotropic model and fixed at 0.74 (C1A–C9) and 0.26 (C1A'–C9') in subsequent refinement cycles.

Data collection: *SMART* (Bruker, 1997–1999); cell refinement: *SAINT* (Bruker, 1997–2003); data reduction: *SAINT*); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

This work was supported by the University of New Brunswick.

References

- Bruker (1997–2003). SAINT. Version 6.45. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1997–1999). SMART. Version 5.054. Bruker AXS Inc., Madison, Wisconsin, USA..
- Decken, A., Rigby, S. S., Girard, L., Bain, A. D. & McGlinchey, M. J. (1997). Organometallics, 16, 1308–1315.
- Johnson, J. W. & Treichel, P. M. (1977). J. Am. Chem. Soc. 99, 1427-1436.
- Nesmeyanov, A. N., Vol'kenau, N. A. & Bolesnova, I. N. (1966). Dokl. Akad.
- Nauk SSSR, 166, 607–615. Pierre, F., Stricker, A., Moinet, C., Sinbandhit, S. & Toupet, L. (1998). J. Organomet. Chem. 553, 253–267.
- Sheldrick, G. M. (1997*a*). *SADABS, SHELXS*97 and *SHELXL*97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Ustynyuk, N. A., Pomazanova, N. A., Novikova, L. N., Kravtsov, D. N. & Ustynyuk, Yu. A. (1986). *Izv. Akad. Nauk SSSR*, **7**, 1688–1689.